M.F.G. Stevens, A. Gescher, J.A. Hickman, K. Vaughan and R.J. Simmonds, Cancer Chemotherapy Group, Department of Pharmacy, University of Aston in Birmingham, Birmingham B4 7ET, UK

Oxidation of the dimethylamino side-chain is believed to play an important role in the metabolic activation of antitumour dimethyltriazenes. When incubated with liver homogenates, or in vivo, these agents are demethylated with the liberation of formaldehyde and the formation of a highly reactive alkylating agent — a monomethyltriazene (Connors & others, 1976). Although the intermediate hydroxymethyltriazenes have hitherto been regarded as only transient species we have achieved a novel synthesis of this type by coupling aryldiazonium salts with a formaldehyde/methylamine mixture in aqueous media. Surprisingly, hydroxymethyltriazenes bearing electron-attracting substituents in the aryl ring are stable crystalline compounds.

The hydroxymethyltriazene (la) proved to be the most active triazene yet tested against the TLX5 lymphoma in mice (Table): it is substantially more active than either the corresponding monomethyltriazene (lb) or the dimethyltriazene (lc).

Table. Antitumour activity of triazenes against the TLX5 lymphoma in mice

$(1) \qquad R - \left\langle \begin{array}{c} - \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - $						
	R	Compor	and R 2	Maximum % increase in survival time	Optimal dose (mg/kg) (5 x daily)	Toxic dose (mg/kg) (5 x daily)
(la)	CO <sub>2</sub> Me	Me	СН2ОН	120	5	160
(lb)	CO <sub>2</sub> Me	Me	Н	87	5	80
(1c)	CO <sub>2</sub> Me	Me	Me	58	40	160*

\*From Connors & others (1976)

In addition the related hydroxymethyltriazene (1:  $R = CO_2Et$ ,  $R_1 = Me$ ,  $R_2 = CH_2OH$ ) displays inhibitory activity in vitro against TLX5 lymphoma cells whereas its dimethyltriazene counterpart (1:  $R = CO_2Et$ ,  $R_1 = R_2 = Me$ ) is inactive. This confirms the activated nature of the former triazene.

Another stable  $\alpha$ -hydroxylated triazene (2a) can be prepared (25% yield) simply by stirring the morpholinotriazene (2b) with potassium permanganate in aqueous acetone at 25°. The morpholin-3-ol (2a) undergoes a 1,2-elimination in acetyl chloride/pyridine at 25° to afford the dehydromorpholinotriazene (3). The morpholin-3-ol (2a) is also detected in extracts from the morpholinotriazene (2b) that have been incubated with rat liver homogenates.

Cl 
$$N=N-N$$
 Cl  $N=N-N$  (3)

Connors, T.A., Goddard, P.M. & others (1976). Biochem. Pharmac., 25, 241-246.